



MICROCOPY RESOLUTION TEST CHART NATIONAL BUREAU OF STANDARDS -1963 - A

OFFICE OF NAVAL RESEARCH

Contract N00014-80-C-0472

Task No. NR 056-749

TECHNICAL REPORT No. 49

Negative-Ion Formation from Surface Scattering and the Anderson Correlation Energy ${\bf U}$

by

Kai-Shue Lam, K. C. Liu and Thomas F. George

Prepared for Publication

in

Physics Letters A

Department of Chemistry University of Rochester Rochester, New York 14627

February 1984

This document has been approved for public release and sale; its distribution is unlimited.



Unclassified
SECURITY CLASSIFICATION OF THIS PAGE (When Date Entered)

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM
I. REPORT NUMBER	2. GOVT ACCESSION	NO. 3. RECIPIENT'S CATALOG NUMBER
UROCHESTER/DC/84/TR-49	AD-A13839	
I. TITLE (and Subtitle)		S. TYPE OF REPORT & PERIOD COVERED
	n from Surface Scattering	İ
and the Anderson Corre	elation Energy U	S. PERFORMING ORG. REPORT NUMBER
		a. PERFORMING ONG. NEW ON I NOMBER
7. AUTHOR(*)		B. CONTRACT OR GRANT NUMBER(#)
Kai-Shue Lam, K. C. Li	iu and Thomas F. George	N00014-80-C-0472
		}
PERFORMING ORGANIZATION NA	ME AND ADDRESS	10. PROGRAM ELEMENT, PROJECT, TASK
Department of Chemisti		NR 056-749
University of Rocheste		NK 050-749
Rochester, New York		
1. CONTROLLING OFFICE NAME AN		12. REPORT DATE
Office of Naval Resear		February 1984
Chemistry Program Code		13. NUMBER OF PAGES
Arlington, Virginia 2	22217 DDRESS(II different from Controlling Office	17 (ce) 15. SECURITY CLASS. (of this report)
4. MONITORING AGENCY NAME & A	ODRESS(II dilibrati from Controlling Circ	Unclassified
		onciassified
		15a. DECLASSIFICATION/DOWNGRADING
		ease and sale; its distribution
is unlimited.		ase and sale; its distribution
This document has beer is unlimited. 7. DISTRIBUTION STATEMENT (of all	n approved for public rele	ase and sale; its distribution
This document has beer is unlimited. 7. DISTRIBUTION STATEMENT (of all	n approved for public rele	ease and sale; its distribution
This document has beer is unlimited. 7. DISTRIBUTION STATEMENT (of all	n approved for public rele	ease and sale; its distribution
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of all all all all all all all all all al	n approved for public rele	ease and sale; its distribution
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of #1 8. SUPPLEMENTARY NOTES Prepared for publicati	n approved for public rele	tase and sale; its distribution (t) (D(+)) (H(-))
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of the supplementary notes Prepared for publication of the supplementary notes Prepared for publication of the supplementary notes.	n approved for public rele	tase and sale; its distribution (trom Report) (+) (D(+)) (H(-)) (D(-)) (ATION ENERGY
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of the state of th	ion Physics Letters A side if necessary and identify by block num CORRE CATTERING H + +	tase and sale; its distribution (+) (D(+)) (H(-)) (D(-)) (ATION ENERGY CESIATED W(100)
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of all of the state of the stat	ion Physics Letters A GCATTERING The approved for public release GOOD	tase and sale; its distribution (+) (D(+)) (H(-)) (D(-)) (ATION ENERGY
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of the supplementary notes Prepared for publication NEGATIVE-ION FORMATION POSITIVE-ION-SURFACE S	ion Physics Letters A GCATTERING The approved for public release GOOD	(+) (D(+)) (H(-)) (D(-)) (ATION ENERGY CESIATED W(100)
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of the state of th	ion Physics Letters A GCATTERING ON-NEWNS MODEL	tase and sale; its distribution (+) (+) (+) (+) (+) (ATION ENERGY CESIATED W(100) AGREEMENT WITH EXPERIMENT
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of #1 8. SUPPLEMENTARY NOTES Prepared for publication on reverse of NEGATIVE-ION FORMATION POSITIVE-ION-SURFACE STWO-ELECTRON TRANSFER TIME-DEPENDENT ANDERSO	ion Physics Letters A SCATTERING ON-NEWNS MODEL	case and sale; its distribution (t) (D(+)) (H(-)) (ATION ENERGY CESIATED W(100) AGREEMENT WITH EXPERIMENT
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of the second	ion Physics Letters A SCATTERING ON-NEWNS MODEL Ide II necessary and identity by block number of the second of	nase and sale; its distribution (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (+) (
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of the supplementary notes Prepared for publication NEGATIVE-ION FORMATION POSITIVE-ION-SURFACE STWO-ELECTRON TRANSFER TIME-DEPENDENT ANDERSO negative-ion formation a unified point of vie correlation energy U in	ion Physics Letters A Side II necessary and identify by block num is CATTERING ON-NEWNS MODEL Ide II necessary and identify by block num is from positive-ion-surfacew. Based on the time-dep	(+) (D(+)) (D(-)) (ATION ENERGY CESIATED W(100) AGREEMENT WITH EXPERIMENT e scattering is presented from endent Anderson-Newns model, the nt role in the two-electron
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of the state of th	ion Physics Letters A SCATTERING ON-NEWNS MODEL Idea II necessary and identify by block number of the good of the probability of the probabilit	case and sale; its distribution (+) (D(+)) (H(-)) (ATION ENERGY CESIATED W(100) AGREEMENT WITH EXPERIMENT e scattering is presented from endent Anderson-Newns model, the nt role in the two-electron ity of negative-jook formation a
This document has been is unlimited. 7. DISTRIBUTION STATEMENT (of the state of th	ion Physics Letters A SCATTERING No. NEWNS MODEL Ide II necessary and identify by block num in from positive-ion-surfacew. Based on the time-dep is seen to play an importal culations of the probabil a experiments on the conve	(+) (D(+)) (D(-)) (ATION ENERGY CESIATED W(100) AGREEMENT WITH EXPERIMENT e scattering is presented from endent Anderson-Newns model, the nt role in the two-electron
This document has been is unlimited. DISTRIBUTION STATEMENT (of the supplementary notes Prepared for publication NEGATIVE-ION FORMATION POSITIVE-ION-SURFACE STWO-ELECTRON TRANSFER TIME-DEPENDENT ANDERSO PROGRATIVE-ION formation a unified point of viecorrelation energy U itransfer process. Cal	ion Physics Letters A SCATTERING No. NEWNS MODEL Ide II necessary and identify by block num in from positive-ion-surfacew. Based on the time-dep is seen to play an importal culations of the probabil a experiments on the conve	case and sale; its distribution (+) (-) (-) (-) (+) (-) (-) (-) (

Physics Letters A, in press

NEGATIVE-ION FORMATION FROM SURFACE SCATTERING AND THE ANDERSON CORRELATION ENERGY U

Kai-Shue Lam*, K. C. Liu and Thomas F. George Department of Chemistry University of Rochester Rochester, New York 14627

ICSU Classification #: 68, 78, 79



Accession For				
NTIS	GRA&I	M		
DTIC TAB				
Unannounced Justification				
	Avail and	/or		
Dist	Special			
	i			
A-1				

*Present address: Department of Physics

California Polytechnic State University

3801 West Temple Avenue Pomona, California 91768

Abstract

A theoretical investigation of negative-ion formation from positive-ion-surface scattering is presented from a unified point of view. Based on the time-dependent Anderson-Newns model, the correlation energy U is seen to play an important role in the two-electron transfer process. Calculations of the probability of negative-ion formation are in good agreement with experiments on the conversion of $H^+(D^+)$ to $H^-(D^-)$ by scattering from a cesiated W(100) surface.

Charge-exchange processes arising from monoenergetic ion beams scattered from solid surfaces have recently been the subject of much experimental $^{1-5}$ and theoretical $^{6-10}$ interest. Most of these studies focus on the process of ion neutralization, involving the transfer of a single electron from the surface to the singly-charged ion. $^{1-3}$, $^{6-10}$ However, owing to the increase of applications involving the transfer of two electrons in a variety of situations, such as plasma fusion 11 , 12 and the generation of high-energy neutral beams, 11 attention has also been directed to processes of negative-ion formation. 4 , 5 , 13

For ion neutralization, the majority of theoretical work has been based on the Anderson-Newns model, 14 originally proposed for the explanation of localized magnetization in transition metal alloys and subsequently applied to various other problems such as chemisorption on metals 15 and mixed valence in solids. 16,17 In these applications of this model, the correlation energy U, arising from the Coulomb repulsion between the two electrons of opposite spin in the same discrete level, plays a crucial role. However, in studies of ion netrualization (or the reverse process of atom ionization), this important quantity is either implicitly assumed to be infinite 6 or completely ignored. $^{7-9}$ Such approaches are thus incapable of accounting for negative ion formation: $U = \infty$ completely supresses the transfer of a second electron, while if only single-electron transfer is considered. U is irrelevant.

Previous theoretical work¹³ on negative-ion formation has centered on the time-dependent width and shift of the valence level of the impact ion, where the time dependence is due to the motion of the ion. It was assumed that first neutralization occurs via a nonresonant Auger process, while the

nuclear motion of the ion can subsequently bring the valence level into resonance with a band state and thus effect the transfer of the second electron. The quantity U again does not play an important role in this theory.

In the present work, we use the Anderson-Newns model to examine the effects of U on the two-electron transfer process in relation to the dynamics of the nuclear motion of the impact ion. This represents the first time that charge-exchange processes in surface scattering have been considered from a unified point of view. We shall see that the negative-ion formation depends crucially on the finite value of U, and in fact, recent experimental results⁵ can be explained in terms of our theory.

The time-dependent Anderson-Newns Hamiltonian including the correlation energy term is given as

$$H(t) = H_0 + V(t), \tag{2}$$

where

$$H_0 = \sum_{\sigma} \varepsilon_{\mathbf{d}} c_{\mathbf{d}\sigma}^{\dagger} c_{\mathbf{d}\sigma} + \sum_{\mathbf{k}\sigma} \varepsilon_{\mathbf{k}} c_{\mathbf{k}\sigma}^{\dagger} c_{\mathbf{k}\sigma} + U n_{\mathbf{d}\sigma} n_{\mathbf{d},-\sigma}$$
 (2)

$$V(t) = \sum_{k\sigma} [V_{kd}(t)c_{d\sigma}^{\dagger}c_{k\sigma} + V_{kd}^{\dagger}(t)c_{k\sigma}^{\dagger}c_{d\sigma}].$$
(3)

The indicies d and k denote the valence state of the impact ion and the conduction band states of the solid, respectively, σ is the spin index and $n_{d\sigma} = c_{d\sigma}^{\dagger} c_{dc}^{\dagger}$. The interaction Hamiltonian, V(t), responsible for the electron transfer between the band and valence states, is the only explicitly time-dependent part. The motion of the projectile ion can be phenomenologically

taken into account by using the specific form⁶

$$V_{kd}(t) = V_{kd}e^{-\lambda|t|}.$$
 (4)

where λ , the sole dynamical parameter in our model, is directly proportional to the normal impact velocity. Hence λ controls the duration of the bound-continuum interaction, while V_{kd} (time-independent) determines its strength.

To lowest order, the perturbative solution for the time-evolution operator \tilde{T} in the interaction picture, which contributes to the S-matrix for negative-ion formation, is given as

$$\tilde{T}^{(2)}(t,-\infty) = (-i)^2 \int_{-\infty}^{t} dt' \tilde{V}(t') \int_{-\infty}^{t} dt'' \tilde{V}(t''), \qquad (5)$$

where

$$\tilde{V}(t) = e^{iH_0t}\tilde{V}(t)e^{-iH_0t}$$
(6)

is V(t) expressed in the interaction picture. At $t=-\infty$, the ion is taken to be infinitely far from the surface, and t=0 is the instant of impact. It is apparent from Eq. (3) that, since negative-ion formation involves ultimately the transfer of two electrons, only terms of even powers in $\tilde{\mathbf{V}}$ in the expansion of $\tilde{\mathbf{T}}$ contribute to the S-matrix.

Figure 1 illustrates schematically the electron-transfer process to lowest, i.e., second, order as contained in Eq. (5). Physically, every factor of $\tilde{V}(t)$ corresponds to the transfer of one electron from the band level ε_k to the ε_d level of the ion, or the conjugate process of electron transfer from ε_d to ε_k . Therefore, the contribution to the S-matrix from

higher-order terms of \tilde{T} can be diagrammatically generated in a straight-forward manner. Such terms (neglected in this work), however, may involve those final states in which the solid becomes electronically excited, with levels higher than the Fermi level occupied. In general, low-ordered processes are favored by high impact velocities (large λ).

To lowest order, the time-dependent probability for negative-ion formation is then given as

$$P(t) = \int_{\epsilon_{1}}^{\epsilon_{F}} d\epsilon \int_{\epsilon_{1}}^{\epsilon_{F}} d\epsilon' \rho(\epsilon) \rho(\epsilon') |\langle k_{x} k_{y}^{\dagger} | \tilde{T}^{(2)}(t,-\infty) | I \rangle|^{2}, \qquad (7)$$

where ε_F and ε_L are the Fermi energy and conduction band edge of the solid, respectively, $\rho(\varepsilon)$ is the density of states of the band, and the initial and final states, |I> and $|k_{\uparrow}k_{\downarrow}^{\dagger}>$ respectively, are described in Fig. 1. The matrix element in Eq. (7) can be evaluated by using Eqs. (2)-(6) to give the result

$$<\mathsf{k}_{+}\mathsf{k}_{+}^{\prime} \mid \tilde{\mathsf{T}}^{(2)}(\mathsf{t},-\infty) \mid \mathsf{I}> = -\frac{v^{2}}{2} \frac{1}{\lambda+\mathsf{i}(\mathsf{U}-\varepsilon-\varepsilon^{\mathsf{i}})/2} \, \mathsf{e}^{\left[2\lambda+\mathsf{i}(\mathsf{U}-\varepsilon-\varepsilon^{\mathsf{i}})\right] \mathsf{t}} \, \left(\frac{1}{\lambda-\mathsf{i}\varepsilon} + \frac{1}{\lambda-\mathsf{i}\varepsilon^{\mathsf{i}}}\right),$$

$$\mathsf{t} \leq \mathsf{0} \qquad (8\mathsf{a})$$

$$= -\frac{v^{2}}{2} \frac{1}{\lambda+\mathsf{i}(\mathsf{U}-\varepsilon-\varepsilon^{\mathsf{i}})/2} \left(\frac{1}{\lambda-\mathsf{i}\varepsilon} + \frac{1}{\lambda-\mathsf{i}\varepsilon^{\mathsf{i}}}\right) - \frac{v^{2}}{2} \frac{1}{\lambda-\mathsf{i}(\mathsf{U}-\varepsilon-\varepsilon^{\mathsf{i}})/2}$$

$$\mathsf{x} \, \left(\frac{1}{\lambda+\mathsf{i}\varepsilon} + \frac{1}{\lambda+\mathsf{i}\varepsilon^{\mathsf{i}}}\right) \left[\mathsf{e}^{-\left[2\lambda+\mathsf{i}(\mathsf{U}-\varepsilon-\varepsilon^{\mathsf{i}})\right] \mathsf{t}} - 1\right)$$

$$+ 2v^{2}\lambda \left\{\frac{1}{(\lambda^{2}+\varepsilon^{2})[\lambda-\mathsf{i}(\mathsf{U}-\varepsilon^{\mathsf{i}})]} \left[\mathsf{e}^{-\left[\lambda-\mathsf{i}(\mathsf{U}-\varepsilon)\mathsf{t}} - 1\right]\right]$$

$$+ \frac{1}{(\lambda^{2}+\varepsilon^{\mathsf{i}/2})[\lambda-\mathsf{i}(\mathsf{U}-\varepsilon)]} \left[\mathsf{e}^{-\left[\lambda-\mathsf{i}(\mathsf{U}-\varepsilon)\mathsf{t}} - 1\right]\right] \qquad \mathsf{t} > \mathsf{0} \qquad (8\mathsf{b})$$

where $\varepsilon = \varepsilon_k - \varepsilon_d$ and $\varepsilon' = \varepsilon_{k'} - \varepsilon_d$ with ε_d set as the zero of energy, and $v = v_{kd}$ is assumed to be independent of energy. As a first approximation, we have also assumed that U and ε_d are constant within the collision region. This latter assumption is not expected to affect the results qualitatively.

We now turn to calculations of the probability P(t) of negative-ion formation and its limiting value at $t \to \infty$, $P(\infty)$, for various choices of the parameters U and λ . $P(\infty)$ represents the experimental observation of outgoing negative ions after the scattering event is complete, and P(t) reflects the behavior of the transient states. The following fixed numerical values were adopted: $\varepsilon_{\parallel} = -10$ eV, $\varepsilon_{\parallel} = -0.06$ eV and $\Delta = \varepsilon V^2 = 0.43$ eV, where we have assumed the band to have no important structure so that the density of states ρ can be taken as constant in the integration over the band in Eq. (7). Although the last two numbers have been used previously in a study of charge transfer in the Na/W(100) system, they have no special significance in our present work, which is to investigate in general the effects of the variation of λ and U, especially the latter.

Intuitively, we expect the probability $P(\infty)$ to be small when the repulsive correlation energy U is large. However, $P(\infty)$ also depends on λ , i.e., on the velocity of the impact ion. As shown in Fig. 2, there is a peak for $P(\infty)$ at a small value of λ for each U. The explanation for this is that since small λ implies a long duration of interaction whereby from the uncertainty principle the resonance requirement is stringent, it is impossible for the second electron in the solid to overcome the barrier U. On the other hand, large λ can ease the resonance requirement -- energy conservation can be violated for short-duration processes -- but it also limits the actual time available for electron transfer, resulting in a small $P(\infty)$. As a consequence, for each U there exists an optimal value of

 λ , λ _m, for which the probability attains a maximum. Moreover, λ _m increases as U is increased, due to the fact that a shorter interaction time is favorable for the second electron to transfer non-resonantly as U becomes larger.

The close relationship between resonant electron transfer and the ion velocity becomes obvious when we look at the time evolution of the probability of negative-ion formation. Figure 3 displays P(t) for various values of U with fixed λ . Each probability curve has a peak at very short time t_m (~ 0.2 femtoseconds). We see that t_m is smaller for larger U, in accordance with the arguments given above. In addition, the degree of transient negative-ion formation, measured by the ratio $P(t_m)/P(\infty)$, is more pronounced for larger U. Figure 4 provides yet another manifestation of the striking transient behavior; namely, although there is an optimal λ_m at each U for the limiting value of the probability, $P(\infty)$, this is not necessarily the case in the transient region.

To test our theory, we shall compare it with measurements on the conversion of $H^+(D^+)$ to $H^-(D^-)$ by scattering from a cesiated W(100) surface at different grazing angles θ . For this purpose, we identify λ as $v\cos\theta$, where v is the magnitude of the velocity, and phenomenologically introduce a velocity-dependent interaction, $\Delta \approx 0.43 \exp(-0.01v)$ (in the unit of eV), to account for the loss of particles due to penetration into the surface. The variation of θ is thus equivalent to the variation of the normal impact velocity λ . Our results, given in Fig. 5, are in qualitative agreement with the experimental ones, Φ where for all incident energies the conversion probability goes

through a maximum. Quantitative comparisons have not been attempted since precise information on critical parameters, especially Δ , is still lacking. For the cesiated W(100) surface, among other complications leading to unreliable data for parameters is the theoretical evidence of a lowering of the work function by multiple dipole formation. 18,19

In this work we have demonstrated, through varying the dynamical conditions of the impact ion, the significance of the correlation energy U in negative-ion formation from positive-ion-surface scattering. Though U in general decreases the probability for regative-ion formation, one can always exploit the experimentally controllable dynamical conditions (varying v and θ) to achieve an optimal result for a given system. Moreover, there may even be the possibility of exploiting the characteristic transient behavior of P(t), since for finite U, P(t_m) is always larger than (∞) except for very large λ . For very small λ , on the other hand, our perturbation approach may not yield correct results, since the long interaction times then allowed may require higher-order processes than the second-order one be considered. Our results have been shown to be in good qualitative agreement with experiments. A more elaborate calculation is needed which takes into account the lowering of the valence level of the ion near the surface 18,19 is needed for quantitative comparision with experiments.

<u>Acknowledgments</u>

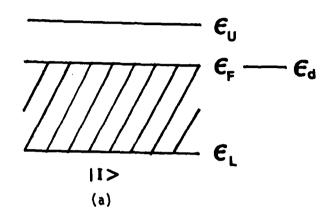
This research was supported in part by the Air Force Office of Scientific Research (AFSC), United States Air Force, under Grant AFOSR-82-0046, and the Office of Naval Research. The United States Government is authorized to reproduce and distribute reprints for governmental purposes notwithstanding any copyright notation hereon. TFG acknowledges the Camille and Henry Dreyfus Foundation for a Teacher-Scholar Award (1975-84) and the John Simon Guggenheim Foundation for a Fellowship (1983-84).

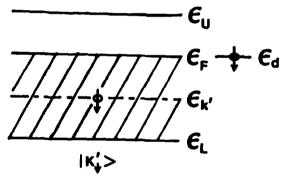
References

- 1. H. H. Brogersma and T. M. Buck, Surf. Sci. <u>53</u>, 649 (1975).
- 2. R. L. Erickson and D. P. Smith, Phys. Rev. Lett. 34, 297 (1975).
- E. G. Overbosch, B. Rasser, A. D. Tenner and J. Los, Surf. Sci. 92, 310 (1980).
- 4. E. G. Overbosch and J. Los, Surf. Sci. <u>108</u>, 117 (1981).
- J. N. M. van Wunnik, B. Rasser and J. Los, Phys. Lett. <u>87A</u>, 288 (1982);
 J. Los, E. A. Overbosch and J. N. M. van Wunnik in <u>Proceedings of the Second International Symposium on the Production and Neutralization of Negative Hydrogen Ions and Beams</u> (Brookhaven, 1980).
- 6. W. Bloss and D. Hone, Surf. Sci. 72, 277 (1978).
- 7. Y. Muda and T. Hanawa, Surf. Sci. 97, 283 (1980).
- 8. R. Brako and D. M. Newns, Surf. Sci. 108, 253 (1981); Vacuum 32, 39 (1982).
- 9. K. L. Sebastian, V. C. Jyothi Bhasu and T. B. Grimley, Surf. Sci. <u>110</u>, L571 (1981).
- 10. J. C. Tully, Phys. Rev. B 16, 4324 (1977).
- 11. K. H. Berkner, R. V. Pyle and J. W. Stearns, Nucl. Fusion 15, 249 (1979).
- 12. K. Wiesemann, K. Prelec and Th. Sluyters, J. Appl. Phys. 48, 2668 (1977).
- B. Rasser, J. N. M. van Wunnik and J. Los, Surf. Sci. 118, 697 (1982).
- 14. P. W. Anderson, Phys. Rev. <u>124</u>, 41 (1961).
- D. M. Edwards and D. M. Newns, Phys. Lett. <u>24A</u>, 236 (1967); D. M. Newns, Phys. Rev. <u>178</u>, 1123 (1969).
- 16. B. Coqblin and J. R. Schrieffer, Phys. Rev. <u>185</u>, 847 (1969).
- 17. J. H. Jefferson and K. W. Stevens, J. Phys. C 9, 2151 (1976).
- 18. K. H. Kingdon and I. Langmuir, Phys. Rev. 21, 380 (1923).
- 19. E. Wimmer, A. J. Freeman, N. Weinert, H. Krakauer, J. R. Hiskes and A. M. Karo, Phys. Rev. Lett. 48, 1128 (1982).

Figure Captions

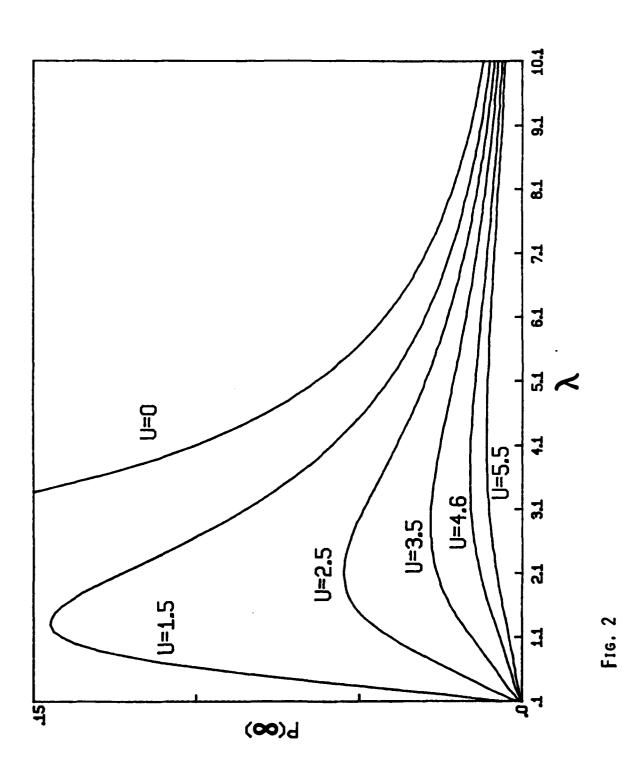
- Fig. 1. Schematic diagram of electron transfer in negative-ion formation. ϵ_U is the upper and ϵ_L the lower edge of the band. (a) Initial state $|I\rangle$: valence state ϵ_d empty, band filled up to the Fermi level ϵ_F . (b) Intermediate states $|k_+\rangle$ and $|k_+^\dagger\rangle$: states corresponding to the neutralized atom; one electron transferred from the ϵ_k or ϵ_{k^\dagger} level to the ϵ_d level. The arrows denote the spin states of the electrons, and the solid and hollow circles represent electrons and holes, respectively. (c) Final states $|k_+k_+^\dagger\rangle$: negative-ion states; two electrons transferred to the ϵ_d level.
- Fig. 2. $P(\infty)$ vs λ for various values of U. Energy is in the unit of eV.
- Fig. 3. P(t) vs t for various values of U with fixed λ . As U increases, the characteristic short-time behavior becomes more pronounced. Energy is in the unit of eV, and time is in the unit of 6.59 x $10^{-16}~{\rm sec}~(\frac{\hslash}{eV})$
- Fig. 4. P(t) vs t for various values of λ with fixed U. The units are the same as in Fig. 3.
- Fig. 5. $P(\infty)$ vs θ , the incident angle of impact. $v_1 < v_2 < v_3 < v_4$.

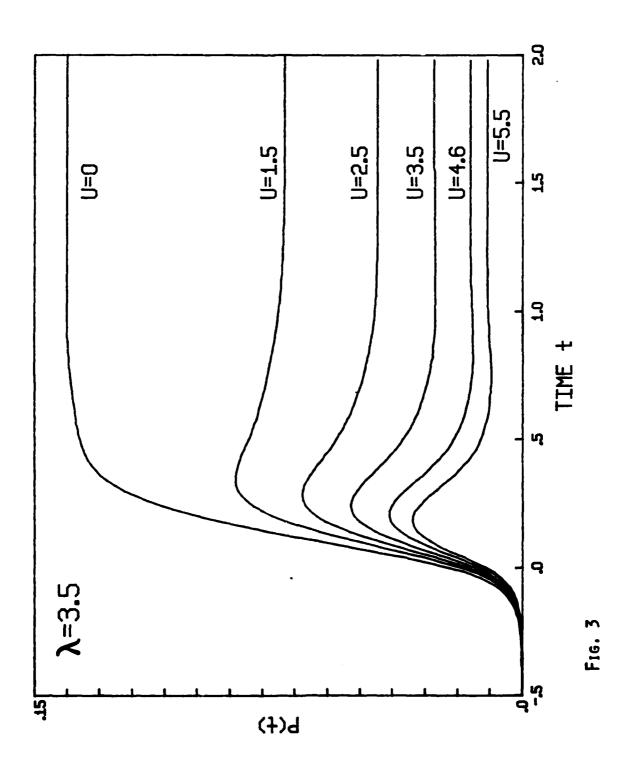


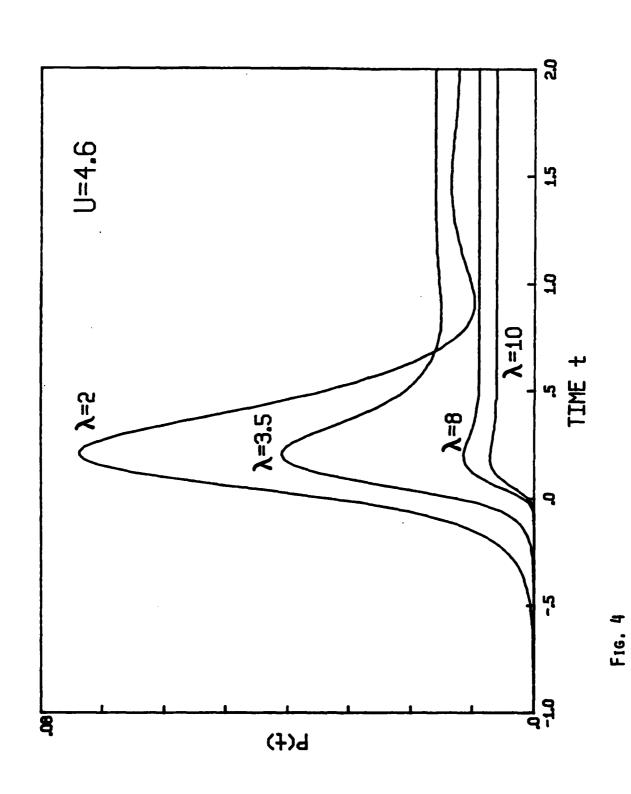


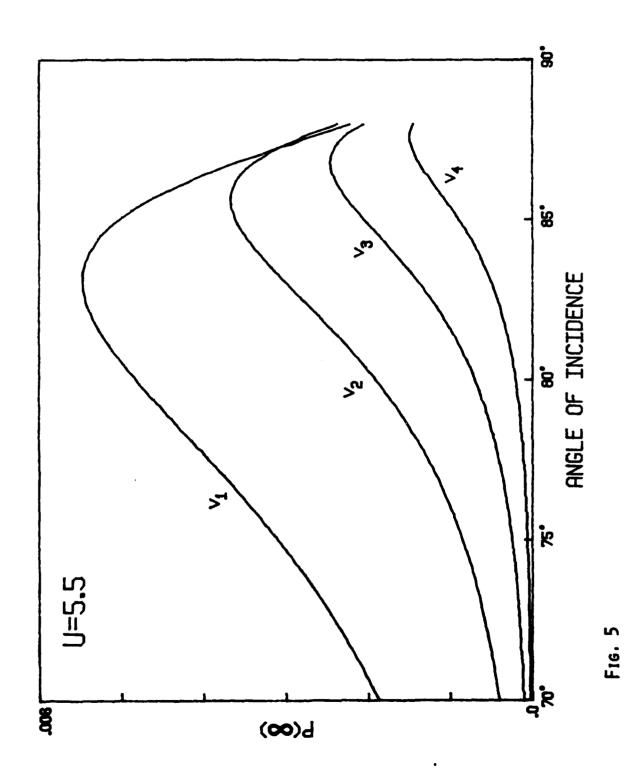
(b)

Fig. 1









DL/413/83/01 GEN/413-2

TECHNICAL REPORT DISTRIBUTION LIST, GEN

<u>c</u>	No. opies		No. <u>Copies</u>
Office of Naval Research Attn: Code 413 800 N. Quincy Street Arlington, Virginia 22217	2	Naval Ocean Systems Center Attn: Technical Library San Diego, California 92152	1
ONR Pasadena Detachment Attn: Dr. R. J. Marcus 1030 East Green Street Pasadena, California 91106	1	Naval Weapons Center Attn: Dr. A. B. Amster Chemistry Division China Lake, California 93555	1
Commander, Naval Air Systems Command Attn: Code 310C (H. Rosenwasser) Washington, D.C. 20360	1	Scientific Advisor Commandant of the Marine Corps Code RD-1 Washington, D.C. 20380	1
Naval Civil Engineering Laboratory Attn: Dr. R. W. Drisko Port Hueneme, California 93401	1	Dean William Tolles Naval Postgraduate School Monterey, California 93940	1
Superintendent Chemistry Division, Code 6100 Naval Research Laboratory Washington, D.C. 20375	1	U.S. Army Research Office Attn: CRD-AA-IP P.O. Box 12211 Research Triangle Park, NC 2770	1
Defense Technical Information Center Building 5, Cameron Station Alexandria, Virginia 22314	12	Mr. Vincent Schaper DTNSRDC Code 2830 Annapolis, Maryland 21402	1
DTNSRDC Attn: Dr. G. Bosmajian Applied Chemistry Division Annapolis, Maryland 21401	1	Mr. John Boyle Materials Branch Naval Ship Engineering Center Philadelphia, Pennsylvania 1911	2
Naval Ocean Systems Center Attn: Dr. S. Yamamoto Marine Sciences Division San Diego, California 91232 Dr. David L. Nelson	1	Mr. A. M. Anzalone Administrative Librarian PLASTEC/ARRADCOM Bldg 3401 Dover, New Jersey 07801	1
Chemistry Program Office of Naval Research 800 North Quincy Street Arlington, Virginia 22217	1		

TECHNICAL REPORT DISTRIBUTION LIST, 056

Dr. G. A. Somorjai Department of Chemistry University of California Berkeley, California 94720

Dr. J. Murday Naval Research Laboratory Surface Chemistry Division (6170) 455 Overlook Avenue, S.W. Washington, D.C. 20375

Dr. J. B. Hudson Materials Division Rensselaer Polytechnic Institute Troy, New York 12181

Or. Theodore E. Madey Surface Chemistry Section Department of Commerce National Bureau of Standards Washington, D.C. 20234

Dr. Chia-wei Woo Department of Physics Northwestern University Evanston, Illinois 60201

Dr. Robert M. Hexter Department of Chemistry University of Minnesota Minneapolis, Minnesota

Dr. J. E. Demuth
I8M Corporation
Thomas J. Watson Research Center
P.O. Box 218
Yorktown Heights, New York 10598

Dr. M. G. Lagally Department of Metallurgical and Mining Engineering University of Wisconsin Madison, Wisconsin 53706

Or. Adolph B. Amster Chemistry Division Naval Weapons Center China Lake, California 93555 Dr. W. Kohn Department of Physics University of California, San Diego La Jolla, California 92037

Dr. R. L. Park Director, Center of Materials Research University of Maryland College Park, Maryland 20742

Dr. W. T. Peria Electrical Engineering Department University of Minnesota Minneapolis, Minnesota 55455

Dr. Keith H. Johnson
Department of Metallurgy and
Materials Science
Massachusetts Institute of Technology
Cambridge, Massachusetts 02139

Dr. J. M. White Department of Chemistry University of Texas Austin, Texas 78712

Dr. R. P. Van Duyne Chemistry Department Northwestern University Evanston, Illinois 60201

Dr. S. Sibener
Department of Chemistry
James Franck Institute
5640 Ellis Avenue
Chicago, Illinois 60637

Dr. Arold Green Quantum Surface Dynamics Branch Code 3817 Naval Weapons Center China Lake, California 93555

Dr. S. L. Bernasek Princeton University Department of Chemistry Princeton, New Jersey 08544

TECHNICAL REPORT DISTRIBUTION LIST, 056

Dr. F. Carter Code 6132 Naval Research Laboratory Washington, D.C. 20375

Dr. Richard Colton Code 6112 Naval Research Laboratory Washington, D.C. 20375

Dr. Dan Pierce National Bureau of Standards Optical Physics Division Washington, D.C. 20234

Professor R. Stanley Williams Department of Chemistry University of California Los Angeles, California 90024

Dr. R. P. Messmer
Materials Characterization Lab.
General Electric Company
Schenectady, New York 22217
12301

Dr. Robert Gomer Department of Chemistry James Franck Institute 5640 Ellis Avenue Chicago, Illinois 60637

Dr. Ronald Lee R301 Naval Surface Weapons Center White Oak Silver Spring, Maryland 20910

Dr. Paul Schoen Code 5570 Naval Research Laboratory Washington, D.C. 20375

Dr. John T. Yates
Department of Chemistry
University of Pittsburgh
Pittsburgh, Pennsylvania 15260

Dr. Richard Greene Code 5230 Naval Research Laboratory Washington, D.C. 20375

Dr. L. Kesmodel
Department of Physics
Indiana University
Bloomington, Indiana 47403

Dr. K. C. Janda California Institute of Technology Division of Chemistry and Chemical Engineering Pasadena, California 91125

Professor E. A. Irene
Department of Chemistry
University of North Carolina
Chapel Hill, Northc Carolina 27514

Dr. Adam Heller Bell Laboratories Murray Hill, New Jersey 07974

Dr. Martin Fleischmann Department of Chemistry Southampton University Southampton \$09 5NH Hampshire, England

Dr. John W. Wilkins Cornell University Laboratory of Atomic and Solid State Physics Ithaca, New York 14853

Dr. Richard Smardzewski Code 6130 Naval Research Laboratory Washington, D.C. 20375

TECHNICAL REPORT DISTRIBUTION LIST, 056

Dr. R. G. Wallis Department of Physics University of California Irvine, California 92664

Dr. D. Ramaker Chemistry Department George Washington University Washington, D.C. 20052

Dr. P. Hansma Physics Department University of California Santa Barbara, California 93106

Or. J. C. Hemminger Chemistry Department University of California Irvine, California 92717

Professor T. F. George Chemistry Department University of Rochester Rochester, New York 14627

Dr. G. Rubloff
IBM
Thomas J. Watson Research Center
P.O. Box 218
Yorktown Heights, New York 10598

Professor Horia Metiu Chemistry Department University of California Santa Barbara, California 93106

Captain Lee Myers AFOSR/NC Bollig AFB Washington, D.C. 20332

Professor Roald Hoffmann Department of Chemistry Cornell University Ithaca, New York 14853 Dr. R. W. Plummer
Department of Physics
University of Pennsylvania
Philadelphia, Pennsylvania 19104

Dr. E. Yeager Department of Chemistry Case Western Reserve University Cleveland, Ohio 41106

Professor D. Hercules University Pittsburgh Chemistry Department Pittsburgh, Pennsylvania 15260

Professor N. Winograd
Department of Chemistry
Pennsylvania State University
University Park, Pennsylvania 16802

Dr. G. D. Stein Mechanical Engineering Department Northwestern University Evanston, Illinois 60201

Professor A. Steckl
Department of Electrical and
Systems Engineering
Rensselaer Polytechnic Institute
Troy, NewYork 12181

Professor G. H. Morrison Department of Chemistry Cornell University Ithaca. New York 14853

Dr. David Squire Army Research Office P.O. Box 12211 Research Triangle Park, NC 27709

